THEORETICAL STUDIES ON ORDER AND PHASE TRANSITIONS IN LIQUID CRYSTALS

Thesis Submitted to the UNIVERSITY OF MYSORE for the Degree of Doctor of Philosophy

Ву

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DECLARATION

I hereby declare that the thesis was composed by me independently and that it has not formed the basis for the award of any Degree, Diploma, Associateship, Fellowship or other similar title.

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CHAPTER I

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INTRODUCTION

Liquid orystals represent states of matter that are intermediate between the crystalline solid and the amorphous liquid. Molecular shape anisotropy is an essential requirement for the existence of such phases. Mesomorphic phases were discovered at the end of the last century by Reinitzer (1888). Several thousands of such compounds have been found since them. The common feature of all such molecules is that they are marrow and relatively long. Recently Chandrasekhar et al (1977) discovered that pure compounds made up of relatively simple molecules of disc-like shape also form mesophases (discotic mesophases).

1.1 Liquid crystals of rod-like molecules

Liquid crystals of rod-like molecules can be formed by the influence of solvents (lyotropic liquid crystals) or by purely thermal processes (thermotropic liquid crystals). In this thesis, we deal with the latter type only. Following the nomenolature proposed originally by Friedel (1922), thermotropic liquid crystals composed of rod-like molecules are classified broadly into three types: nematic, cholesteric and smectic.

Nematic liquid orystals are the simplest type of liquid crystals having a degree of long range orientational order (denoted by 3) of the molecules, but no long range translational order. The molecules tend to be parallel to some common axis referred to as the director (see fig.1).

If the molecules forming the nematic phase are optically active, then they exhibit a twisted mematic or cholesteric phase in which the director follows a helical arrangement (fig. 2). The degree of twist is characterized by the pitch of the helix. At the gross level, the thermodynamics of cholesteric and nematic phases are almost similar as the energy associated with the twist is $\sim 10^5$ times smaller than the total energy associated with the parallel alignment of molecules (Saupe, 1968). However, in the molecular statistical theories of cholesteric phases, the nature of the molecules suit be specifically taken into account. Keating (1969), Goossens (1971), Priest and Labensky (1974), wan der Meer et al (1976a,b), Lin Lei et al (1977) among others have developed detailed molecular statistical theories of the cholesteric phases.



FIGURE 1: Schematic representation of molecular order in the nematic phase.



tal: FIGURE 2: The cholesteric liquid cry6 Schematic representation of the helical structure.

The smootic phase has both orientational and some translational order. From a structural point of view, all smootics have layered arrangement, with a well defined interlayer spacing. The smootics are sub-classified as smectic A, smootic B, etc. depending upon the molecular arrangement in the layer. For smootic A, the average direction of orientation of molecules is normal to the plane of the layer. The centres of gravity of molecules show no long range translational order within the layer and each layer is a two dimensional liquid (see fig. 3). The optic axis is thus normal to the plane of the layers and the system is optically uniaxial. We shall not be discussing other types of smootics here.

Many compounds exhibit both the smeatic A and nematic phases and in such cases the nematic phase usually occurs at temperatures above those corresponding to the smeatic phase. For example 4-(4-n-butyloxybenzylidenephenyl)azoaniline (Arora and Fergason, 1971) shows the following transitions:

 131.0°C
 135.0°C
 184.5°C

 solid
 ← → accatio
 → isotropic

However Cladis (1975) discovered that there may be a different sequence of transitions in some cases. She found that binary mixtures of certain mesogenic dyano

_____ _____ _____. _____ _____

smectic A structure. Schematic representation of ... M^ FIGURE

PIGURE 4

(a) Schematic representation of a 'liquid-like' stack of disc-shaped molecules (from Levelut, 1980). (b) Hexagonal modification D_h of an upright columnar structure. The filled circle depicts the molecular core which is normal to the columnar axis. (c) Rectangular modification D, of an upright columnar structure. The circular cores are normal to the columnar ares and form a face-centered rectangular lattice, but the asymmetric disposition of the chains results in a herringbone pattern of ellipses, whose ellipticity is highly exaggerated here for the sake of clarity (Tinh et al, 1981; Destrade et al, 1981 and Levelut, 1980). (d) Tilted columnar structure D. (Tinh et al, 1981 and Destrade et al, 1981). (e) Nace-centered rectangular lattice of D. The cores are tilted with respect to the columnar axes (Tinh et al, 1981 and Destrade et al, 1981). (f) Nematic phase N p of disc-shaped molecules (Tinh et al. 1981 and Destrade et al. 1981).

compounds exhibit the following sequence of phase transition on cooling:

isotropio --- nematio --- smectic A --- nematio

The second nematic phase, which occurs at temperatures lower than those of the smeotic phase is called the 're-entrant mematic' phase in analogy with similar 're-entrant' phases observed in superconductors, He-3, etc. Madhusudana et al (1979), Hardouin et al (1979a) and Tinh and Gasparoux (1979) have subsequently observed re-entrant phenomena in some pure components. For some of the compounds a re-entrant smectic phase which exists at temperatures lower than those of the re-entrant mematic phase has also been found (see Hardouin et al, 1979b).

1.2 Liquid crystals of disc-like molecules

Disc-shaped molecules have been recently shown (Chandrasekhar et al, 1977) to exhibit a new class of liquid crystalline phases (the discotic phases). A number of disc-shaped molecules are now known to exhibit thermotropic mesomorphism (see Chandrasekhar, 1982). The mesophases so far discovered fall into two distinct categories, the columnar and the menatic. In columnar (D) type, the discs are stacked one on top of the other, the different



(c)

(f)

columnar stacks forming a two-dimensionally periodic array. Several variants of this structure have been identified; hexagonal, rectangular, tilted, etc. (see fig. 4a-4e). The mematic $(B_{\rm D})$ is an orientationally ordered arrangement of the discs with no long range translational order (see fig. 4f). The twisted mematic (cholesteric) phase was observed in a family of disc-like molecules (see Destrade st al, 1980a). It has also been found that some disc-like molecules show a re-entrant nematic phase (Destrade et al, 1980b).

1.3 Properties of mematic liquid crystals

In several chapters of this thesis, we shall be discussing the statistical thermodynamics of the nematic phase of rod-like molecules. Hence, we shall first summarize briefly the properties of nematic liquid crystals of such molecules.

The nomatic phase exhibited by molecules of arbitrary shaps can be characterised by the orientational order parameters

$$S_{ij}^{\alpha\beta} = \frac{1}{2} \left\langle 31_{\alpha} j_{\beta} - \delta_{\alpha\beta} \delta_{ij} \right\rangle \begin{cases} \alpha, \beta = 1, 1, 2 \\ \\ i, j = \xi, \eta, \zeta \end{cases}$$
(1)

where , Y, Z) refer to the (orthogonal) laboratory frame

and (ξ, η, ζ) to the frame linked to the molecule. δ_{ij} and $\delta_{\alpha\beta}$ are Kronecker deltas. $S_{ij}^{\alpha\beta}$ is a symmetric tensor in ij and $\alpha\beta$, and has zero trace. By taking the director \hat{n} along the 2-axis, we get

$$S_{ij} = \frac{1}{2} \langle 3i_{g} j_{g} - \delta_{ij} \rangle \qquad (2)$$

where

$$S_{\xi\xi} = \langle \frac{1}{2} (3\sin^2 \theta \cos^2 \gamma - 1) \rangle$$

$$S_{\xi\eta} = \langle -\frac{3}{2}\sin^2 \theta \cos \gamma \sin \gamma \rangle$$

$$S_{\xi\eta} = \langle -\frac{3}{2}\sin \theta \cos \theta \cos \gamma \rangle$$

$$S_{\eta\eta} = \langle \frac{1}{2} (3\sin^2 \theta \sin^2 \gamma - 1) \rangle$$

$$S_{\eta\varsigma} = \langle \frac{3}{2}\sin \theta \cos \theta \sin \gamma \rangle$$

$$S_{\varsigma\varsigma} = \langle \frac{1}{2} (3\cos^2 \theta - 1) \rangle$$
(3)

Here Θ is the angle made by the molecular axis (ζ -axis) with the Z-axis and γ is the Eulerian angle made by the line of interaction of XY and ξ_{η} planes with the η -axis (see fig. 5). Since the medium is apolar, all odd terms in Θ and γ should vanish, i.e.,

3. is now a diagonal matrix with sero trace. Hence for a uniaxial nematic medium made of biaxial particles, there



FIG.5 The Eulerian angles defining the orientations of (ξ, η, ζ) with reference to (X, Y, Z).

are two independent order parameters (see Freiser, 1970; Alben et al, 1972; Alben, 1973; Straley, 1974; Luckhurst et al, 1974; and Gelbart and Barboy, 1979). These are defined as

$$S = S_{\zeta\zeta} = \iint \frac{1}{2} (\Im \cos^2 \theta - 1) f(\theta, \gamma) d(\cos \theta) d\gamma$$

$$D = S_{\zeta\zeta} - S_{\eta\eta} = \iint \frac{3}{2} \sin^2 \theta \cos 2\gamma f(\theta, \gamma) d(\cos \theta) d\gamma \quad (4)$$

where $f(\theta, \gamma)$ is the single particle distribution function. S gives the degree of orientational order of the long axes of the molecules with respect to the mematic director. The order parameter D describes the difference between the tendencies of the two transverse molecular axes to align along the mematic director, i.e., D arises due to the deviation from axial symmetry around the long molecular axis .

If the molecules are cylindrically symmetric, the distribution function will be independent of γ . This simplifies $3_{\xi\xi}$ and $3_{\eta\eta}$ to

$$S_{\xi\xi} = S_{\eta\eta} = \frac{1}{2} \left< \frac{3}{2} \sin^2 \Theta - \frac{1}{2} \right>$$
$$= -\frac{3}{2}$$
$$1 \ 0 \ . = 0 \ .$$

The matrix Sij will then be

$$S_{1j} = \begin{bmatrix} -5/2 & 0 & 0 \\ 0 & -5/2 & 0 \\ 0 & 0 & 5 \end{bmatrix}$$

The only independent order parameter S is given by

$$S = \int \frac{1}{2} (3 \cos^2 \theta - 1) f(\theta) d(\cos \theta)$$
 (5)

The distribution function $f(\theta)$ is symmetric about $\theta = \pi/2$, i.e., $f(\theta) = f(\pi - \theta)$. The general appearance of $f(\theta)$ is given in fig. 6. The order parameter S takes a maximum values equal to 1 when all the molecules are packed parallel to the director. And it takes a value equal to zero, when the molecules can take all possible orientations with equal probability.

The order parameter S is a function of temperature. It decreases as the temperature increases (see fig. 7) and drops abruptly to zero at the nomatio-is stropic transition temperature $(T_{\rm NI})$. The NI transition is of first order, associated with a heat of transition (Δ H) which is typically only a few per cent of that of the crystal-nematio transition. Since the transition is weak, it is associated with strong pre- and post-transition effects. For instance



FIGURE 7: Experimental curves of S for p-azoxyanisole (PAA), anisaldazine and p-azoxyphenetole (PAP) (Reproduced from Chandrasekhar et al, 1970 and Madhusudana et al, 1971). magnetic (Δn_{H}) and electric birefringence (Δn_{H}) of the isotropic phase exhibit strong anomalies as the temperature is lowered to T_{NI} (Tsvetkov and Ryuntsev, 1968). Similar anomalies are seen in the flow birefringence (Tolstoi and Fedotov, 1947) and nuclear spin lattice relaxation (Cabane and Clark, 1970).

Associated with the variation of the order parameter, the specific heats C_p and C_v , the isothermal compressibility β and the coefficient of thermal expansion α increase sharply as the temperature approaches $T_{\rm NI}$ in the nematic phase (see figs. 8, 9, 10).

High pressure studies on menatic liquid crystals show that the order parameter S at $T_{\rm NI}$ is very nearly independent of the pressure. Further the curves of constant order parameter in $\ln V - \ln T$ diagram are almost parallel (see fig. 11). The slope of these lines is given by

$$\gamma = -\left[\frac{\partial \ln T}{\partial \ln V}\right]$$
(6)

γ is a measure of the relative importance of volume V compared to that of temperature T in determining the variation of order parameter S of the medium. For menatic

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FIGURE 8: Experimental curve of specific heat at constant volume C_v in the nematic and isotropic phases for PAA derived from the observed values of C_p (Arnold, 1964) and β (Kapustin and Bykova, 1966).



FIGURE 9: Experimental curve of isothermal compressibility β in the mematic and isotropic phases for PAA (from Kapustin and Bykova, 1956).



FIGURE 10: Experimental curve for the coefficient of thermal expansion α in the nematic and isotropic phases for PAA (from Maier and Saupe, 1960).



FIG. 11: Experimental curves of ln V vs. ln T for values of constant nematic order parameter S for FAA (from McColl and Shih, 1972).

liquid crystals of rod-like molecules, y ranges from 4 to 6. (see McColl and Shih, 1973; Horn, 1978; Kuss, 1978; Keyes and Daniels, 1979).

The first theory of the mematic phase was proposed by Born in 1916 and since then several attempts have been made to develop a suitable statistical model. As is well known, there is no exact statistical theory of simple liquids made up of spherical molecules. For the mematic, which is an oriented liquid composed of highly anisotropic molecules, the theory becomes even more complicated.

1.4 Dope of the thesis

this thesis, we present our contributions to the study of statistical thermodynamic properties of nematic phase exhibited by <u>rod-like</u> and <u>disc-like</u> molecules. Further, in the case of discotic liquid crystals we have also presented some calculations on the more ordered columnar and smectic A phases.

In chapter II, we have discussed the statistical theory of nematic phase based on attractive interactions alone. In particular the short range order effects in the nematic and isotropic phases are studied. The mean field theory developed by Maier and Saupe (1958-1960) and its

subsequent improvements (Chandrasekhar and Madhusudana, 1971; Humphries et al, 1972) are successful in predicting the qualitative behaviour of the thermodynamic properties of nematic liquid crystals. However, they predict too strong the a NI transition. If TH is hypothetical second order transition point, the mean field theory predicts that $(T_{\rm HI}-T^{\rm st})\simeq$ 40 K whereas the emerimental value is $\sim 1^{\circ}$ K. Further the theory fails to explain the short range order effects in the nematic and isotropic phases. In an attempt to overcome some of these drawbacks, Madhusudana and Chandrasekhar (1973a) and Ypma and Vertogen (1976) developed a theory which includes short range correlations between neighbouring molecules, employing the Bethe cluster model. In Bethe's model, a molecule is assumed to have z mearest neighbours and its interaction with one of its nearest neighbours is taken as

$$\mathbb{E}(\Theta_{ij}) = -\mathbb{B}^{\mathbb{H}}\mathbb{P}_{2}(\cos \Theta_{ij})$$

where θ_{ij} is the angle between the long axes of the molecules i and j. The outer molecule of the Bethe cluster is subjected to a mean field potential $V(\theta_j)$ due to the rest of the medium. Several approximations, namely, (1) Chang's consistency condition, (ii) Krieger-James (KJ) approximation, (iii) Bethe-Peierls-Weiss (BPW) approximation, and (iv) constant coupling (CO) approximation have been proposed

for solving Bethe's cluster model. We have examined these approximations at some length and emphasized that satisfying the KJ relation as accurately as possible is essential to achieve thermodynamic consistency. For this purpose, the long range mean field potential $\forall(\Theta_j)$ of an outer molecule j is taken as

$$\mathbb{V}(\Theta_j) = \sum_{n=0}^{\infty} B_{2n} P_{2n} (\cos \Theta_j)$$

For a given value of \mathbf{z} , various properties of the menatic and isotropic phases and $\mathbf{T}^{\mathbf{H}}$ are calculated by retaining terms up to $\mathbf{F}_2(\cos \theta_j)$, $\mathbf{F}_4(\cos \theta_j)$ and $\mathbf{F}_{12}(\cos \theta_j)$ in the expansion for $\mathbf{V}(\theta_j)$. The calculations are repeated for $\mathbf{z} = 3$, 4, 8 and 12.

We have found that the error in satisfying KJ relation is extremely small when terms up to $P_{12}(\cos \theta_j)$ are included.i: In this case, the ratio $(2_{\rm HI} - 2^{\rm H})/2_{\rm HI}$ decreases from 0.117 to 0.031 as z decreases from co (mean field) to 3. The value of a given long range order parameter $\langle P_{2n} \rangle$ Is lower for lower values of z particularly near $2_{\rm HI}$. Further the rate of decrease of $\langle P_{2n} \rangle$ with temperature increases for lower values of z. The cawlated values of $\langle P_2 \rangle \equiv \langle P_4 \rangle$ and $\langle P_4 \rangle / \langle P_2 \rangle$ for a = 3are in good agreement with the experimental data. We

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observe that C_{ψ} (specific heat at constant volume) at as we as ΔC_{ψ} increase as we go from the mean field limit $(z = \infty)_{to} = 4$ in agreement with expectations. At temperatures sufficiently for off from $T_{\rm HI}$, C_{ψ} is lower if z is lower.

The part played by permanent dipoles in determining the nematic liquid crystalline properties has been studied in chapter III. There are a large number of menatogenic compounds with a mitrile (-CEN) group fixed to one end of the molecule. Since the menatic director is apolar, there is an equal probability for the dipoles to point in either direction. Further for strongly polar molecules, the interaction energy between meighbouring dipoles is comparable with the dispersion energy. Madhusudana and Chandrasekhar (1973b) showed that near neighbours of such strongly polar molecules will fixed an antiparallel arrangement. However the absence of long range translational order in menatic liquid crystals precludes the possibility of antiferroelectric long range order.

We have refined Batha's cluster model developed for a system of polar molecules (Madhusudana and Chandrasekhar, 1973c) in the light of results obtained for non-polar molecules. $E(\Theta_{14})$ and $V(\Theta_{4})$ are taken as

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$$\mathbb{E}(\Theta_{ij}) = A^{H}P_{1}(\cos \Theta_{ij}) - B^{H}P_{2}(\cos \Theta_{ij})$$
$$\mathbb{V}(\Theta_{i}) = -B_{2}P_{2}(\cos \Theta_{j}) - B_{4}P_{4}(\cos \Theta_{j})$$

We have made numerical calculations by assuming that $A^{H}/B^{H} = 0.5$. For z = 3, 4 and 8, various properties of the mematic and isotropic phases and T^{H} are calculated. The dielectric constants (ε_{11} and ε_{\perp}) along and perpendicular to the unique axis of the mematic phase are also calculated for the same values of z.

The variation of $(T_{\rm HT} - T^{\rm M})/T_{\rm HT}$ is found to be similar to that observed for the system of non-polar molecules. The values of $\langle P_2 \rangle$ of a typical cyanobiphenyl compound (Miyano, 1978) agrees well with the theoretical values for s = 3 and 4. However, the experimental $\langle P_4 \rangle$ values of such compounds are lower than the theoretical values even for s = 3. As in the system of non-polar molecules, 0, and ΔQ , at $T_{\rm NI}$ increase systematically as g decreases from op to 4. The calculated dielectric anisotropy is comparable to the experimental value. ÁT) interesting consequence of the theory fa that the mean dielectric constant should increase by a few per cent on going from the nematic to the isotropic phase because of the discontinuous decrease in value of $\langle \Gamma_1(\cos \theta_{ij}) \rangle$ =

This is found to be the case experimentally in a number of strongly positive materials.

Because of strong antiparallel correlations between near neighbours, now it is known that cyano compounds form a bilayer structure. This has some important consequences like the occurrence of re-entrant menatic phase, smectic A_1 smectic A_2 transition, etc.

In the above theories the anisotropy of the molecular shape which is an important factor in determining whether or not a substance exhibits liquid crystalline phase is not taken into account. In other words the short range intermolecular repulsions are not considered. In chapter IV we give a summary of different hard rod theories proposed so far. We have also discussed hybrid models which in addition to the repulsive force, also take into account the long range anisotropic interaction between molecules. 01 them Cotter's model (1977) developed on the basis of the scaled particle theory (SPT) for a system of spherocylinders gives a comparatively better description of the nematic phase. SPT was originally developed by Reiss et al (1959) for a system of hard spheres. The equation of state of a system of hard spheres given by SPT is identical to the compressibility equation of state obtained by the Percus-

Yevick (P-Y) theory. The theory provides a means of deriving approximate expressions for the chemical potential μ and pressure P of a hard particle fluid by considering the reversible work necessary to insert a scaled particle (i.e., a 'solute' particle which is a scaled replica of the 'solvent' particles) at some arbitrary point in the fluid. The chemical potential of a spherical particle of radius ca is written as (Reiss et al, 1959; Cotter and Hartire, 1970a,b,c)

$$\mu/kT = \ln(\rho \Lambda^{2}) + \pi(q, \rho)/kT$$
 (7)

where \wedge^3 is the reciprocal of the particle translational momentum partition function, $\rho = N/V$ is the number density, $W(\xi, \eta)$ is the reversible work necessary to orgate a spherical cavity of radius at least equal to $\xi = (\alpha + 1)a$ into a fluid of N spherical molecules of radius a. It is related to the probability $P_{\alpha}(\xi, \rho)$ of observing a fluctuation in which such a cavity can be found, where

$$P_{0}(\xi, r) = \exp[-i(\xi, r)/kt]$$
 (8)

The chemical potential of the solvent particle is obtained by setting $\xi = 2a$ in eq.(7). The pressure of the system is given by

$$\frac{P}{kT} = 1 - \frac{a}{5} \left[\frac{\ln F_0(q, p)}{\partial q} \right]$$
(9)

Cotter and Martire (1970a,b,c) showed that when α is sufficiently small so that three 'regular' spheres cannot simultaneously be in contact with the scaled sphere, P_o can be written as

$$P_0(\xi, \xi) = 1 - \frac{4}{3}\pi a^3 \epsilon \left[1 + 3\alpha + 3\alpha^2 + \alpha^3\right].$$
 (10)

In the other extreme when a is very large, W must approach the reversible PV work of creating a macroscopic spherical cavity in the liquid, i.e.,

$$\lim_{\alpha \to \infty} \left[u \right] = \frac{4}{3} \pi a^3 \alpha^3 P . \qquad (11)$$

This suggests that $\ln P_0(\xi, \gamma)$ can be expanded as - $\ln P_0(\xi, \gamma) = C_0 + C_1 \alpha + C_2 \alpha^2 + (\frac{4}{3}\pi a^3 p) \alpha^3; \alpha \ge 0$. The coefficients C_{μ} are determined from eq. (10) together with

$$\sigma_{m} = \frac{-1}{m!} \left[\frac{\partial^{m} \ln \mathcal{F}_{0}(\boldsymbol{\xi},\boldsymbol{\rho})}{\partial \boldsymbol{\alpha}^{m}} \right]_{\boldsymbol{\alpha}} = 0 \qquad (12)$$

The pressure P is then obtained by using equation (9) together with equation (12) to get

$$\frac{P \Psi_0}{kT} = \frac{d + d^2 + d^2}{(1 - d)^3} ; \quad d = \frac{4}{3} \pi a_S^3. \quad (13)$$

The Gibbs free energy and the Helpholts free energy are given by

$$\mu/kT = G_{c}/NkT = ln \circ + W/kT , \qquad (14)$$

$$A_{MKT} = G_{MKT} - P/PKT .$$
(15)

As we have already montioned, equation (IS) is nothing but the P-Y equation of state of hard spheres. This gives exact second and third virial coefficients and reasonable values for the higher virial coefficients also.

Cotter (1970a,b,c) has extended this theory to a system of hard spherocylinders. The reversible work W written in terms of the volume excluded to the centre of a regular spherocylinder with orientation k by the presence of the scaled spherocylinder which is the orientation 1, is a function of $|\sin \Theta_{ik}|$. By retaining terms up to $P_2(\cos \Theta)$ in the expansion of $|\sin \Theta|$, Cotter made the calculation for $x (= 1.4 \ 1/2r) = 3$ by treating the hard rod repulsive interactions in terms of SPT and subjecting each nolecule to a mean field attractive potential of the form

 $\Psi_{1} = - \vartheta_{0} \rho - \vartheta_{2} \rho \langle P_{2} \rangle P_{2}(\cos \theta_{1})$

where ϑ_0 and ϑ_2 are the strengths of the isotropic and anisotropic interactions.

We have extended the calculations for various values of x for a system of hard spherocylinders as well as for a system with a superposed attractive potential. By comparing the various transition properties obtained for a system of hard spherocylinders with the experimental data on PAA, we have found that $x \simeq 2.5$ gives a good overall agreement. Including terms up to $P_{12}(\cos \theta)$ in $|\sin \theta|$ expansion, there is a considerable increase in the values of $\langle P_2 \rangle$, density change $\Delta P/\bar{\rho}$ and entropy change $\Delta S/Nk$ at NI transition. Properties obtained for $x \simeq 2.5$ are again in relatively better agreement with the experimental results than for other values of x.

Including the attractive potential Ψ_1 , we have made calculations by fixing d (packing fraction) = 0.62 at $\Xi_{\rm HI} = 409^{\circ}$ K. Truncating |sin θ | expansion at $P_2(\cos \theta)$ various transition properties and $\gamma = -[\partial \ln \pi/\partial \ln V]_{\langle P_2 \rangle}$ calculated for different values of x are compared with the experimental results on PAA. γ which is equal to 1 when x = 1 increases as x increases, first slowly and then rapidly. It attains the experimental value ~ 4 for x $\simeq 1.75$. The order parameter $\langle P_2 \rangle$ and $\Delta P/\bar{\rho}$ increase with x as expected. $\Delta S/Nk$ is not much affected by the value of x and is of the order of 0.5 which is considerably larger than the experimental value (~ 0.17). G_p decreases with x whereas a and β do not vary much. Calculations beyond x~2.45 are not possible if we choose to retain the correct values for

d and T_{HI} . For $x \simeq 1.75$, many properties are in an overall agreement with the experimental results. Retaining terms up to $P_{12}(\cos \theta)$ in the $|\sin \theta|$ expansion, there is a substantial increase in the values of $\langle P_2 \rangle$, $\Delta P|\overline{\rho}$ and $\Delta s/Mk$. On comparison with the experimental date. A value of x lying between 1.2 and 1.5 would appear to give a relatively better agreement than for other values of x.

The theory also enables us to study the thermodynamic properties of mematic and isotropic phases at the NI transition as functions of pressure. By retaining terms up to $P_{12}(\cos \theta)$ in the $|\sin \theta|$ expansion, we have calculated phase transition properties for x = 1.75 by varying the pressure from one bar to 6.5 kbar. The results are compared with the experimental results on PAA (see for example Chandrasekhar and Shashidhar, 1979) (dr/dP)°K/kbar ~ 30 up to ikbar, whereas the experimental value ~ 47 . df/dP decreases as Int increases, which is in qualitative agreement with the experimental trend noticed by Stishov et al for PAA and Horn (1978) for methoxybensylidene-butylaniline (MDBA) and 4-n-pentyl-4'-cyanobiphenyl (50B). Variation of molar volume of the nematic phase (V_n) as a function of \mathbb{Z}_{NI} is of the same order as that of experimental variation. $\langle P_{o} \rangle$ is practically constant, varying slowly from 0.499 to 0.511

when $T_{\rm HI}$ changes from 409° to 572°K. T the alope of the curve obtained by plotting ln V versus ln $T_{\rm HI}$ varies from 4.0 to 6.5 as $T_{\rm HI}$ increases from 409°K to 572°K. Apart from this, the theory is able to reproduce the qualitative behaviour of $\Delta 3$, internal energy change (ΔU), molar volume of nematic ($V_{\rm R}$) and isotropic ($V_{\rm I}$) phases, the change in molar volume (ΔV), etc., as functions of pressure.

The isotropic fluid state of hard spherocylinders has also been studied with the aid of computers (Vieillard Baron, 1974; Monson and Rigby, 1978; Carnahan and Starling, 1969). The results of SFT in the isotropic phase for x = 1, 2 and 3 are in good agreement with the results of computer studies for low densities. However SFT overestimates the pressure as the density is increased and the discrepancy also increases with increase of density.

It is of interest to extrapolate the essentially 'exact' results of the computer studies to the region of nematic-isotropic phase transition. We found that a model proposed by Andrews (1975) for calculating the equation of state of an assembly of hard spheres can be extended to the case of spherocylinders. Further the extended model provides a suitable scheme for making calculations in the ordered phase also. We will present here a brief summary of Andrews

approach developed for a system of spheres. The activity of a fluid of spheres of diameter σ is computed as a product of two terms. The first term is the probability that a point \mathbf{x} chosen at random within a volume \mathbf{V} does not lie inside the core of one of the N-molecules. The second term is the probability conditional on the first, that the hole around the given point can actually accommodate the introduced molecule. This demands that there should be no other molecule within the additional volume $\frac{4}{3}\pi\sigma^3 - \frac{1}{6}\pi\sigma^3 = \frac{7}{6}\pi\sigma^3$. If ω gives the value of the average volume taken up by each molecule when they are crowded together, then the reciprocal activity can be written as

$$a^{-1} = (1 - \frac{1}{6} \pi \sigma^3 \rho) \exp[-\frac{7\pi \sigma^3 \rho}{6(1 - \omega \rho)}]$$
 (16)

Andrews assumed that $\omega = \omega_0 + \omega_1 \rho$. At low densities, the value of ω_0 is obtained from the third virial coefficient of a dilute fluid of spheres. In the other extreme, when the density is very large, the crystal clearly has the close-packed volume of $\omega_{\text{solid}} = 1/\rho_0 = \sigma^3/2$. Under these two conditions, the coefficients ω_0 and ω_1 are calculated. The simplified expression for the pressure of the system of spheres will be

$$\frac{p}{\rho kT} = \frac{5.1834 \ z}{(1 - 1.534 \ z + 0.534 \ z^2)} - \frac{1.3504}{z} \ln (1 - 0.7405 \ z)$$
$$- \frac{4.8534}{2} \ln (1 - 1.534 \ z + 0.534 \ z^2)$$
$$- \frac{4.8534}{2} \ln (1 - 1.534 \ z + 0.534 \ z^2)$$
$$- \frac{15.977}{z} \ln (\frac{1 - 0.534 \ z}{1 - z})$$
(17)

where $Z = \frac{P}{\rho_0}$ is the scaled density relative to the close-packed density. Compared to the results of SPT, the values of P/PkT obtained from equation (17) are in better agreement with the molecular dynamics data of Alder and Wainwright (1960). Further unlike in SPT, the unphysical calculations beyond $P > \rho_0$ are not admissible since P/PkT (see eq. 17) tends to take infinite value as P approaches ρ_0 .

We extended the Andrews approach to a system of spherocylinders. We found that the two term expansion of ω is inadequate to reproduce the value of P/PkT obtained from the computer calculations. Hence we expanded ω as $\omega = \sum_{n=0}^{6} \omega_n e^n$. The coefficients $\omega_0 * \omega_1 * \cdots * \omega_6$ are derived by using e_0 of a close packed system of spherocylinders and the first eight virial coefficients got by Monte Carle calculations of Honson and Rigby (1978).

As in SPT, we have made calculations for a system

of hard spherocylinders as well as for a system which includes an attractive potential, by retaining terms up to $P_2(\cos \theta)$, and $P_{12}(\cos \theta)$ in the $|\sin \theta|$ expansion (see chapter V). On comparing with the results of SPT, we have found that the present calculations improve upon the values of x as well as other results, particularly the second derivatives of the obenical potential.

As we have already noted, disc-like molecules also exhibit liquid crystalline phases which can be broadly classified as columnar and nomatic phases. Transition between columnar, nematic and isotropic phases have been observed in a number of discotic mesogens.

In chapter VI, we extend the SPT to study properties at the nematio-isotropic transition of disc-like mesogens. In order to be able to compare the results on disc-like molecules with those on rod-like molecules, the shape of the molecule is taken to be a right dircular cylinder. If r is the radius of the cylinder and 1 its height, the shape factor R = r/l > 0.5 for disc-like molecules and < 0.5 for rod-like molecules. The excluded volume in this case involves three different functions of Θ_{ik} , vis., [sin Θ_{ik}], $|\cos \Theta_{ik}|$ and $\int_{0}^{\pi/2} \sqrt{1 - \sin^2 \Theta_{ik}} \sin^2 \varphi \, d\varphi$. By retaining terms up to $P_2(\cos \Theta_{ik})$, $P_4(\cos \Theta_{ik})$ and $P_6(\cos \Theta_{ik})$ in

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their expansions, properties of rod-like and disc-like particles are studied for a system of hard particles as well as far a system superposed with an attractive potential. The results obtained are compared with the experimental data. Our calculations demonstrate that it is necessary to retain higher order terms, at least up to $P_4(\cos \theta)$ in the expansion of angle dependent terms in order to get qualitatively correct results. The phase transition properties of a menatic medium of disc-like molecules are essentially analogous to those of rod-like molecules, when they are plotted as functions of the ratio (larger dimension/smaller dimension) of the cylinder.

In chapter VII, we have extended the McMillan's (1971) model of smootic A phase to the columnar phase exhibited by disc-like molecules. As we have already noted, in the smectic A phase the molecules are upright within the layer. in each layer, with their centres irregularly spaced. The director (say Z-axis) also represents the direction of the layer normal (or the vector of the density wave). McMillan has proposed a simple description of this phase by extending the Maier-Saupe theory to include an additional order parameter for characterizing the one-dimensional translational periodicity of the layered structure. To start with he has considered a pair potential of the form

$$V_{12}(r_{12}, \cos \theta_{12}) = -\frac{V_0}{Nr_0^3 \pi^{3/2}} \exp(-(r_{12}/r_0)^2)P_2(\cos \theta_{12})$$

where r_{12} in the distance between the molecular centres and θ_{12} the angle between the symmetry axes of the molecules 1 and 2. r_0 is of the order of the length of the rigid part of the molecule and V_0 the strength of the anisotropic interaction. By retaining the leading terms in the Fourier expansion, the single particle potential in the mean field approximation turns out to be

$$V_1(2, \cos \theta) = -V_0[\langle P_2 \rangle + a \cos(2\pi 2/d)]P_2(\cos \theta)$$

where $\alpha = 2\exp[-(\pi r_0/d)^2]$ is the model parameter.

We have extended this mean field model to describe the columnar phase exhibited by disc-like molecules by treating the density wave to be periodic in two dimensions. The theory is developed for a face-centered rectangular (FOR) lattice composed of liquid-like columns, taking the molecular cores to be circular discs normal to the columnar axes. The FOR lattice is described by a superposition of three density waves with wave vectors say \vec{A} , \vec{B} and $\vec{C} = \vec{A} + \vec{B}$ defined as in fig. 12. Unlike in the smectic A phase of rod-like molecules, the director is parallel to the columnar axis (2-axis say), whereas the wave vectors \vec{A} , \vec{B} and \vec{C} are





FIG.12: Two-dimensional face-centered rectangular lattice showing (a) the lattice parameters a and b, and the primitive cell of the direct lattice, (b) the reciprocal lattice vectors \vec{A} , \vec{B} , and \vec{C} .

in the orthogonal II plane. Therefore, we couple each density wave to the appropriate component of the orientational order parameter along I, B and C, rather than to the component along Z. The single particle potential in the mean field approximation is written in terms of a_1 , a_2 and order parameters η_1 , η_2 , σ_1 and σ_2 , by retaining only the leading terms in the Fourier expansion of the density wave. a, is an interaction strength related to the density wave along \vec{d} and \boldsymbol{a}_2 that along \vec{A} and \vec{B} . η_2 and η_1 are the orientational order parameters measured along the X and Y axes, and σ_2 and σ_1 are the order parameters coupling the orientational and translational order along the \vec{A} (or \vec{B}) and \vec{C} directions respectively. The single particle distribution function and the free energy are written in the mean field approximation. These equations give rise to four possible solutions corresponding to biaxial rectangular columnar, biaxial szectio, uniaxial nematic and isotropic phases.

We have studied the phase diagram as well as the transition entropies for $\frac{b}{a} = 1.5$, 1.6, $\sqrt{3}$, 1.85 and 1.95. For $\frac{b}{a} = \sqrt{3}$, the phase diagram is exactly identical to the one presented by Feldkamp et al (1981) for hexagonal lattice. The hexagonal-mematic transition is always first order. The

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phase diagrams for values of $\frac{b}{a} < \sqrt{3}$ are very similar to that for the hexagonal lattice and lead to a first order transition between a, biaxial rectangular columnar phase and a uniaxial nematic phase. The temperature range of the nematic phase decreases with increasing a_1 and for values of a_1 higher than some value say a_0 , the columnar phase transforms directly to the isotropic phase. If as in McMillan's theory, a_1 is interpreted to be a measure of the chain length, these results are in qualitative accord with the experimental results.

If $\frac{b}{a}$ is slightly greater than $\sqrt{3}$ say equal to 1.85, the phase diagram is again rather similar to that for the hexagonal lattice. However as the asymmetry of the lattice is increased, we gat; solutions corresponding to a smectic phase also. For $\frac{b}{a} = 1.95$ we find that for $\alpha_1 < 0.52$, the rectangular columnar phase transforms to a biaxial smectic 4 phase (with the layer normal along \vec{d}) which in turn undergoes a second order transition to a uniaxial nomatic phase at a higher temperature.

By comparing the phase diagrams for $1.5 \le \frac{b}{a} \le 1.95$, we find that the value of a_0 increases as $\frac{b}{a}$ increases. The temperature variation of order parameters in different phases show that larger the asymmetry of the lattice,

stronger is the biaxial order of the rectangular columnar phase.

Some of the results discussed in this thesis are reported in the following publications.

- 1 Short range orientational order in mematic liquid crystals (N.V.Nadhusudana, K.L.Savithranna and S. Chandrasekhar) <u>Pramans</u> 8, 22 (1977).
- 2 The mematic-isotropic phase transition: Application of the Andrews method (K.L.Savithramma and H.V.Madhusudana) <u>Kol. Cryst. Liq. Cryst.</u> 62, 63 (1980).
- 3 Scaled particle theory of a system of hard right circular cylinders (K.L. Savithramma and N.V. Madhusudana) presented at the Righth International Liquid Crystal Conference, Kyoto, July 1980 - Mol. Cryst.Liquid Cryst. <u>74</u>, 245 (1981).
- 4 Extension of McMillan's model to liquid crystals of disc-like molecules (S. Chandrasekhar, K.L.Savithramma and N.V.Madhusudana) - Presented at the Fourth International Symposium on Liquid Crystals and Ordered Fluids, Las Vegas, March 28-April 2, 1982 (Plenum Press)
- 5 Scaled particle theory of a system of right circular cylinders subjected to an attractive potential (X.L. Savithranna and N.V. Madhusudana) - <u>Mol. Cryst.</u> <u>Liq. Cryst.</u> ((in press). 1982 MCLC G0 35-45
- 6 Extension of the scaled particle theory calculations on a system of spherocylinders to high pressures 1983 (K.L.Savithrassa and N.V.Hadhusudana) (under preparation).

MCLC 97 407-415

The following paper on magnetism has some discussions relevant to the chapter II of the thesis.

7 Equivalence of the Krieger-James approximation and the constant-coupling approximation in magneticm (K.L. Savithramma and H.V. Madhusudama) - <u>Promana</u> 10, 349 (1978).

REFERENCES

- Alben, R. 1973 Phys. Bev. Lett. 30, 778.
- Alben, R., McColl, J.R. and Shih, C.S. 1972 Sol. State Commun. <u>11</u>, 1081.
- Alder, B.J. and Wain Wright, T.E. 1960 J. Chem. Phys. 23, 1439 Andrews, P.C. 1975 J. Chem. Phys. <u>62</u>, 272
- Arnold, H. 1964 Z. Phys. Chem. (Leipzig) 226, 146.
- Arora, S.L. and Fergason, J.L. 1971 Chen. Soc. Faraday Div. Symposium, Nr. 5, 97.
- Born, M. 1916 Sits d. Phys. Math. 25, 614
- Brooks, J.D. and Taylor, G.H. 1965 Carbon 3, 185.
- Carnehan, N.F. and Straling, K.E. 1969 J. Chem. Phys. <u>51</u>, 635. Cabane, B. and Clark, W.G. 1970 Phys. Rev. Lett. <u>25</u>, 91.
- Chandrasekhar, S. 1982 Advances in Liquid Crystals, Vol.5 (in press).
- Chandrasekhar, S. and Madhusudana, N.V. 1971 Acta Cryst. <u>A27</u>, 303.
- Chandrasekhar, S., Sadashiwa, B.K. and Suresh, K.A. 1977 Prasana <u>9</u>, 471.
- Chandrasekhar, S. and Shachidhar, R. 1979 Advances in Liquid Crystals, 4, 83.
- Chandrasekhar, S., Shashidhar, R. and Tara, N. 1970 Mol. Cryst. Liquid Cryst. <u>10</u>, 337.

- Cladis, P. Z. 1975 Phys. Rev. Lett. 35, 48.
- Cotter, M.A. 1977a J.Chem. Phys. 66, 1098
- Cotter, M.A. 1977b Mol. Cryst. and Lig. Cryst. 39, 175.
- Cotter, M.A. and Martire, D.E. 1970a J.Chem. Phys. 52, 1902.
- Cotter, M.A. and Martire, D.E. 1970b J. Chem. Phys. 52, 1909.
- Cottor, M.A. and Martire, D.E. 1970c J. Chem. Phys. 53, 4500.
- Destrade, C., Malthete, J., Tinh,N.H., and Gasparoux, H. 1980b Phys. Lett. <u>784</u>, 82.
- Destrade, C., Tinh, K.H., Malthete, J. and Jacques, J. 1980a Phys. Lett. <u>794</u>, 189.
- Friedel, G. 1922 Ann. Physique 8, 275.
- Friedel, E. 1925 C.R. Acad. Sci. 180, 269.
- Preiser, M.J. 1970 Phys. Bev. Lett. 24, 1041.
- Gelbart, W.M. and Barboy, B. 1979 Mol. Oryst. Lig. Oryst. 55, 209
- Goossens, M.J.A. 1971 Nol. Cryst. Lig. Cryst. 12, 237.
- Hardouin, F., Sigaud, G., Achard, M.F. and Gasparoux, H. 1979a Phys. Lett. 714, 347.
- Hardouin, F., Sigaud, G., Achard, M.F. and Gasparoux, H. 1979b Sol. State Commun. <u>30</u>, 265.
- Norm, 3. G. 1978 J. de Physique, <u>39</u>, 167.
- Humphries, R.L., James, P.G. and Luckhurst, G.R. 1972 J. Chem. Soc. Faraday Trans. 2, <u>68</u>, 1031.

LIBRARY RAMAN RESEARCH BANGALOR CLASS NO. 043:538.9 LC SAV ACT NO. 23225 .

- Kapustin, A.P. and Bykova, H.T. 1966 Sov. Phys. Crystallogr. 11, 297.
- Keating, P.N. 1969 Mol. Oryst. Liq. Cryst. 8, 315.
- Keyes P.H. and Daniels, W.B. 1979 J.de Physique 40, 380.
- Kuns, E. 1978 Mol. Gryst. and Liq. Gryst. 47, 71.
- Lin-Liu, Y.R., Shih, Y.M. and Woo, C.W. 1977 Phys. Rev. <u>A15</u>, 2550.
- Luckhurst, G.R., Zannoni, C., Nordio, P.L. and Segre, U. 1975 Mol. Phys. 30, 1345.
- Madhusudana, N.V. and Chandrasekhar, S. 1973a Solid State Comma 13. 377.
- Madhusudana, N.V. and Chandrasekhar, S. 1973b, Pramina. 1, 12
- Madhusudana, N.V. and Chandrasekhar, S. 1973c Proceedings of the International Liquid Crystals Conference, Bangalore, Pramana Supplement 1, 57.
- Madhusudana, N.V., Sadashiwa, B.K. and Moodithaye, K.F.L. 1979 Current Science 48, 613.
- Nadhusudana, H.V., Shashidhar, R. and Chandrasekhar, S. 1971 Mol. Cryst. Liq. Cryst. 13, 61.
- Maler, W. and Saupe, A. 1958 Z.Naturforsch. 13a, 564.
- Maier, W. and Saupe, A. 1959 Z.Naturforsch. 14a, 882.
- Maler, W. and Saupe, A. 1960 Z. Naturforsch. 15a, 287.

- McColl, J.R. and Shih, C.S. 1972 Phys. Rev. Lett. 29, 85
- McHillan, W.L. 1971 Phys. Rev. A4, 1238.
- Merret, W.G., Cole, G.D. and Walker, W.W. 1971 Mol. Cryst. Idg. Cryst. <u>15</u>, 105.
- Miyano, X. 1978 J. Chem. Phys. <u>69</u>, 4807.
- Konson, P.A. and Rigby, M. 1978 Nol. Phys. 35, 1337.
- Onsager, L. 1949 Ann. N.Y. Acad. Sol. 51, 627
- Priest, R.G. and Lubensky, T.C. 1974 Phys. Rev. A9, 893.
- Reinitzer, F. 1868 Montasch Chem. 9, 421.
- Reiss, H., Frisch, H.L. and Lebowitz, J.L. 1959 J.Chem. Phys. 31, 369.
- Saupe, A. 1963 Angew-Chem. Int. Edn. 7, 97.
- Stishov, S.H., Ivanov, V.A. and Kachinskii, V.N. 1976 JETP Lett. 24, 297.
- Straley, J.P. 1974 Phys. Rev. A10, 1881.
- Tinh,N.H., Destrade, C. and Gasparoux, H. 1979 Phys. Lett. <u>724</u>, 251.
- Tinh,N.H. and Gasparoux,H. 1979 Mol. Cryst. Liq. Cryst. 49, 281.
- Tolstoi, N.A. and Fedotov, L.B. 1947 J. Exp. Theor. Phys. (USSR) 17, 564.
- Tswetkov, V.N. and Ryuntsev, E.I. 1968 Sov. Phys. Crystallogr. 13, 225.

Vieillard-Baron, J. 1974 Mol. Phys. 28, 809.

van der Meer, B.M. and Vertogen, G. 1976 Phys. Lett. A59, 279.

- wan der Meer, B.M., Vertogen, G., Dokker, A.J. and Ypma, J.G.J. 1976 J. Chem. Phys. <u>65</u>, 3935.
- Ypma, J.G.J. and Vertogen, G. 1976 Sol. State Comm. 18, 475.